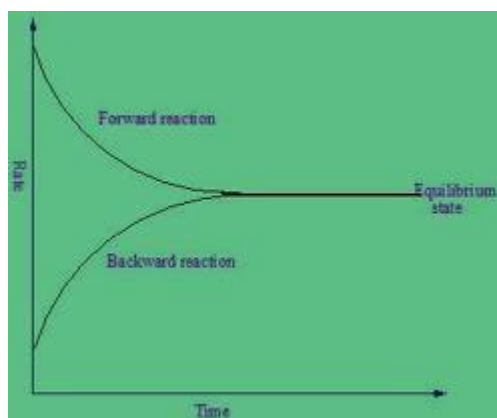


EQUILIBRIUM

Reversible reaction: A reaction that can proceed in both directions. Or a reaction that can both go forward and backward.

Dynamic equilibrium:



Closed system:

Le Chatelier's principle:

Effect of changing the pressure on position of equilibrium:

Effect of changing temperature on the position of equilibrium:

Effect of changing the concentration on the position of equilibrium:

Effect of using a catalyst on the position of equilibrium:

Manufacturing conditions:

Equilibrium constant:

Calculating the equilibrium constant:

Effect of temperature on equilibrium constant:

Left or right?

